SYNTHESIS OF 2,3,4,5-TETRAHYDROPYRIMIDO[2,1-b]-BENZOTHIAZOLES

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1-Thiocarbamoyl-2-phenylpyrazolidines are cyclized to give 2,3,4,5-tetrahydropyrimido[2,-1-b]benzothiazoles, the structure of which was proved by mass spectrometry.

It is known that arylthiosemicarbazides on heating in acidic media are capable of cyclizing to 2-aminobenzothiazoles [1, 2]. We have been able to show that three-ring 2,3,4,5-tetrahydropyrimido[2,1-b]-benzothiazoles (I) can be similarly synthesized from 1-thiocarbamoyl-2-phenylpyrazolidines. The reaction apparently proceeds via a Fischer prototropic rearrangement, but in our case the aromatic ring attacks the sulfur atom rather than the carbon atom.

$$\begin{array}{c|c}
S & NH_2 \\
R & R
\end{array}$$

$$\begin{array}{c|c}
H & S & NH_3 \\
\hline
N: & NH_4 \\
R & R
\end{array}$$

$$\begin{array}{c|c}
R & 1a,b
\end{array}$$

Ia R=H; b R~CH₃

The mass spectrum of Ia confirms the assigned structure:

$$[M]^{+}$$
 $C_{2}H_{4}$ N_{+} N_{-} N_{+} N_{-} N_{-}

The formation of compounds of the I type is simultaneously a proof that an amino group rather than a hydrazine nitrogen atom is eliminated in the process.

EXPERIMENTAL

2,3,4,5-Tetrahydropyrimido[2,1-b]benzothiazole (Ia). A mixture of 3.8 g (0.02 mole) of 1-phenyl-pyrazolidine hydrochloride and 4 g (0.04 mole) of potassium isothiocyanate was heated in 10 ml of absolute alcohol on a water bath for 12 h, after which it was cooled, and the precipitate was removed by filtration, washed repeatedly with ice water, and recrystallized from alcohol to give 2.5 g (58%) of 1-thiocar-bamoyl-2-phenylpyrazolidine with mp 165°. Found: C 58.4; H 6.2%. $C_{10}H_{13}N_{3}S$. Calculated: C 58.2; H 6.3%. UV spectrum (in ethanol): λ_{max} 250 nm (log ϵ 4.30). IR spectrum (in mineral oil): 3200-3400 cm⁻¹.

A solution of 1.2 g (0.006 mole) of 1-thiocarbamoyl-2-phenylpyrazolidine in 5 ml of absolute methanol saturated at 0° with hydrogen chloride was heated in a sealed ampule at 130° for 6 h, after which the mixture was evaporated, and the residue was chromatographed on a column filled with Al_2O_3 with successive elution with benzene and benzene-chloroform (2:1) to give 0.7 g (63%) of Ia with mp 123° (from octane). Found: C 63.0; H 5.4%. $C_{10}H_{10}N_2S$. Calculated: C 63.2; H 5.3%. UV spectrum (in ethanol): λ_{max} , nm, (log ϵ): 225 (4.62), 268 (4.00), 298 (3.73). IR spectrum (in mineral oil): 1615 cm⁻¹.

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3-Methyl-2,3,4,5-tetrahydropyrimido[2,1-b]benzothiazole (Ib). A mixture of 1.3 g (0.007 mole) of 1-phenyl-4-methylpyrazolidine hydrochloride and 1.25 g (0.013 mole) of potassium isothiocyanate in 4 ml of absolute alcohol was heated on a water bath for 10 h, after which it was worked up as in the preceding experiment to give 1.43 g (99%) of 1-thiocarbamoyl-2-phenyl-4-methylpyrazolidine with mp 143° (from alcohol). Found: C 60.2; H 6.8%. $C_{14}H_{15}N_{3}S$. Calculated: C 59.8; H 6.8%.

The cyclization was carried out as in the preceding experiment with 1.2 g of 1-thiocarbamoyl-2-phenyl-4-methylpyrazolidine. The eluate was evaporated, and the residue was vacuum-distilled to give 0.61 g (55%) of Ib with bp 204° (1 mm) and mp 72° (from octane). Found: C 65.2; H 6.3%. $C_{11}H_{12}N_2S$. Calculated: C 64.9; H 5.9%. UV spectrum (in ethanol), λ_{max} , nm (log ϵ): 225 (4.61), 268 (4.05), 297 (3.79). IR spectrum (in mineral oil): 1640 cm⁻¹.

The mass spectrum was recorded with an MKh-1303 spectrometer with admission of the sample into the ion source.

LITERATURE CITED

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